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## **Skill Practice 30 #'s 1-5**

Skill Practice 1 Name: \_\_\_\_\_ Date: \_\_\_\_\_  
Hour: \_\_\_\_\_ For problems 1-3, please use these conversion factors: 1 pallet = 45 bundles 1 bundle = 32 cases 1 case = 12 cans 1 can = 218.4 mL 1. How many

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mL of apple juice does the company need to make to fulfill an order for 2.5 pallets? 9,434,880 mL 2. If 46,680 mL of juice are produced, how many ...

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How many moles of Iron (Fe) are present in a sample containing  $4.41 \times 10^{22}$  atoms? moles How many molecules of Oxygen ( $O_2$ ) are present in a sample that is  $1.83 \times 10^{-23}$  moles? molecules How many moles of Hydrogen ( $H_2$ ) are present in a sample weighs 8.57 g? moles

### **Mass and Mole Relations Exercises**

over and over again, chemists refer to it as a mole (abbreviated mol). 6.022  $\times 10^{23}$  objects is called a mole, just as you call

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12 objects a dozen. Look again at how these quantities are related. 55.847 g of iron 6.022 10<sup>23</sup> iron atoms 1 mol of iron 32.066 g of sulfur 6.022 10<sup>23</sup> sulfur atoms 1 mol of sulfur General Plan for Converting Mass, Amount,

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### **CHEMQUEST 31 USING MOLES WITH FORMULAS ANSWERS PDF**

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Name: Date: Hour: I. Calculate the number of molecules in 210 grams of 'water.  $7.02 \times 10^2$  molecules 2. If you have  $6.25 \times 10^2$  molecules of sulfur tetrafluoride, how many grams do you have?

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Skill Practice 36 Name: \_\_\_\_\_ Date: \_\_\_\_\_  
Hour: \_\_\_\_\_ 1. Under water where the temperature is 17°C and the pressure is 394 kPa, a diver inhales 2.1 L of air from his SCUBA tank. a) How many moles of gas are in his lungs? 0.343 mol b) If the diver swims to the surface without exhaling where the temperature is 32°C and the

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